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Jul 6, 1999

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TITLE: Solid phase extraction of phenethyl alcohol

BSPR:

The production of aroma compounds represents an important tool in the production of natural flavors for food and beverage industries. E. Albertazzi, et al., Biotech. Lett. 16, 491-6 (1994). One of the most commonly used aroma compounds is phenethyl alcohol. Phenethyl alcohol (e.g., phenethanol, 2-phenylethanol, or benzeneethanol) is naturally present in many essential oils, and has a rose-like/honey floral odor. Phenethyl alcohol is chemically synthesized or extracted in commercial applications as a flavoring or fragrance for consumer goods, such as perfumes and food.

ity or alkalinity. In t at which the solu-7) is not always the

portance in a large such as water purifor leather, in preselectroplating baths, ous other instances.

ingicide formulations iophthalimide.

tark for cyclobarbital

a standardized mix-1,2,3,3,5-hexamethyl-

with musk-like odor 5°C and becomes liqconventional solvents

or a pharmaceutical

r a pharmaceutical

conditions in which liquid, or gas(vapor). ily on the concentrasolids are the most ilds occupy the internormally crystalline, ses are without struc-

listinct, and mechanispersion or solution. quid, or gaseous (vann the major comporexternal phase and ispersed or internal not be uniformly discovered colloid chemis-

'illard Gibbs (q.v.) in al system F = n-r-2 rmodynamic equilibre actants. The number red in a given heteroied by analysis or obuph by proper choice ses (r), and the indesemperature and pres-

rule apply to all mulsolvent blends, glass,

nine.

-methyl-1,5-cyclohex-CH₃)₂JCH:CH. A

is (a) d- and (b) l-op-

ble in water; soluble °C); b.p. 66-68°C (16 (b) Sp. gr. 0.8324 m); refractive index

Derivation: (a) Found in ginger oil; Ceylon and Seychelles cinnamon oil. (b) Found in eucalyptus oil. Hazard: Moderately toxic by ingestion and skin absorption, strong irritant. Uses: Flavoring; perfumery.

 $\label{eq:chi} beta-\textbf{phell} \\ \tilde{\textbf{andrene}} \quad (4\text{-isopropyl-1-methylene-2-cyclohexene}) \quad CH_2: CCH: CHCH[CH(CH_3)_2] CH_2: CH_2: CHCH[CH(CH_3)_2] CH_2: CHCH[CH(CH(CH_3)_2] CH_2: CHCH[CH(CH(CH_3)_2] CH_2: CHCH[CH(CH(CH_3)_2]$

nocyclic terpene occurring as (a) d- and (b) l-optical isomers.

Properties: (a) Mobile oil with pleasant odor and a burning taste. Sp. gr. 0.8520 (20°C); b.p. 171-172°C (760 mm); refractive index 1.4788. (b) Mobile oil; sp. gr. 0.8497 (15°C); b.p. 178-179°; flash point 120°F (T.C.C.). Toxicity unknown; refractive index 1.4800. Both are insoluble in water and alcohol; soluble in ether.

Derivation: (a) lemon oil. (b) Japanese peppermint oil

Hazard: Flammable, moderate fire risk.

"Phemerol."330 Trademark for benzethonium chloride (q.v.).

phenacaine hydrochloride

C₂H₅OC_nH₄NCH(CH₃)NC₆H₄OC₂H₅·HCl·H₂O. N,N'-Bis(para-ethoxyphenyl) acetamidine hydrochloride.

Properties: Small, white crystals; odorless; faintly bitter taste. Incompatible with alkalies. M.p. 190°C. Soluble in alcohol, boiling water and chloroform; less so in cold water, insoluble in ether.

Grades: N.F.; technical.

Use: Medicine

phenacemide (phenylacetylurea) C₆H₅CH₂CONHCONH₂.

Properties: White to creamy white, odorless, tasteless crystalline solid; m.p. 212-216°C; slightly soluble in alcohol, benzene, chloroform and ether; very slightly soluble in water.

Use: Medicine

phenacetin. U.S.P. name for acetophenetidin (q.v.). phenacyl chloride. See chloroacetophenone.

phenacyl fluoride. See fluoroacetophenone.

"Phenamine."307 Trademark for a series of direct dyestuffs, used for the dyeing of cotton and paper.

phenanthraquinone. See phenanthrenequinone.

phenanthrene C₁₄H₁₀. A tricyclic hydrocarbon. Properties Colorless, shining crystals. Soluble in alcohol, ether, benzene, carbon disulfide and acetic acid; insoluble in water. Sp. gr. 1.063; m.p. 100.35°C, b.p. 340°C. Combustible.Derivation: Fractional distillation of high-boiling coal-tar oils, with subsequent recrystallization from alcohol.

Hazard: Photosensitizes skin, and may be a carcinogen.
Uses: Dyestuffs; explosives; synthesis of drugs;

biochemical research; phenanthrenequinone.

phenanthrene acetamide. C₁₆H₁₃NO. A carcinogen.

phenanthrenequinone. (Erroneously: phenanthraquinone) $C_{14}H_8O_2$.

Properties: Yellow-orange, needle-like crystals. Soluble in sulfuric acid, benzene; glacial acetic acid and hot alcohol; slightly soluble in ether; insoluble in

water, Sp. gr. 1.4045; m.p. 206-207°C; b.p. sublimes above $360^{\circ}\mathrm{C}_{\odot}$

Derivation: By oxidation of a boiling solution of phenanthrene in glacial acetic acid with chromic acid, solution in sodium disulfite, precipitation by means of hydrochloric acid and recrystallization.

Uses: Organic synthesis; dyes.

1,10-phenanthroline (4,5-phenanthroline; ortho-phenanthroline) C₁₂H₈N₂·H₂O. A heterotricyclic compound.

Properties: White crystalline powder; m.p. 93-94°C, anhydrous 117°C. Slightly soluble in water; soluble in alcohol, benzene.

Derivation: Made by heating ortho-phenylenediamine with glycerin, nitrobenzene and concentrated sulfuric acid; or in like manner from 8-aminoquinoline. Uses: Forms a complex compound with ferrous ions used as an indicator; drier in coatings industry.

phenarsazine chloride. See diphenylaminechloroarsine.
phenazine (azophenylene) C₆H₄N₂C₆H₄. A tricyclic compound.

Properties: Yellow crystals; m.p. 170-171°C; b.p. > 360°C; very slightly soluble in water; soluble in alcohol and ether. Combustible.

Hazard: Probably toxic.

Uses: Organic synthesis; manufacture of dyes; larvicide.

phenethicillin. See potassium alpha-phenoxyethyl penicillin.

phenethyl acetate. See 2-phenylethyl acetate.

phenethyl alcohol (phenylethyl alcohol; 2 phenylethanol; benzyl carbinol) C₆H₅CH₂CH₂OH.

Properties: Colorless liquid; floral odor; sharp burning taste; sp. gr. 1.017-1.020 (25°C); refractive index (n 20/D) 1.5310-1.5340; m.p. -27°C; b.p. 219°C. Flash point 216°F. Soluble in 50% alcohol; soluble 1 part in 50 parts of water; soluble in fixed oils, alcohol, and glycerol; slightly soluble in mineral oil. Combustible.

Derivation: (a) By reduction of phenylacetic ethyl ester by sodium in absolute alcohol. (b) By the action of ethylene oxide on phenylmagnesium bromide and subsequent hydrolysis.

Grades: Technical; N.F.; F.C.C.

Containers: Tin cans and glass bottles; drums. Uses: Organic synthesis; synthetic rose oil; soaps; flavors; antibacterial; preservative; medicine.

Hazard: May be highly toxic by inhalation or ingestion.

sec-phenethyl alcohol. See alpha-methylbenzyl alcohol. phenethylamine. See 2-phenylethylamine.

phenethyl anthranilate. See 2-phenylethyl anthranilate.

phenethyl isobutyrate. See 2-phenylethyl isobutyrate.
phenethyl phenylacetate. See 2-phenylethyl phenylacetate.

phenethyl propionate. See 2-phenylethyl propionate. phenethyl salicylate. See 2-phenylethyl salicylate.

ortho-phenetidine (2-aminophenetole) NH₂C₆H₄OC₂H₅. Properties: Oily liquid; rapidly becomes brown on exposure to light or air. Solidifies below -20°C; b.p. 228-230°C. Soluble in alcohol and ether; insoluble in water. Combustible.

Superior numbers refer to Manufacturers of Trade Mark Products. For page number see Contents.

fenuron.

aminobenzene)

unstable in air; usually hydrochloride; sp. gr 287°C; soluble in alco-

a-dinitrobenzene or nirochloric acid Purified

purity

n. MCA warning label. detection of nitrous ; laboratory reagent. Other restricted arti-

ine: ortho-diaminoben-

nic crystals; darkens in).p. 252-258°C; soluble chloroform: somewhat

o-dinitrobenzene or nirochloric acid. Purified

purity. hotographic developing ratory reagent.

ninobenzene)

de crystals (oxidizes on nd black); m.p. about alcohol, ether; soluble 1 by light. Flash point

i-dinitrobenzene or niochloric acid. Purified

estion and inhalation; ince, 0.1 mg per cubic label

notographic developing 1 measurements: intertioxidants and accelerreagent.

Other restricted arti-White label.

-(meta-hydroxyphenylıloride)

hite crystals; odorless; acid to litmus paper; in alcohol; m.p. 140-

alcohol.

.0970 (20/20°C); b.p., 01 mm (20°C); wt 9.1 viscosity 1.01 poise .c.). Combustible.

uffs. ie.

2-phenylethyl acetate (phenethyl acetate)

C₆H₅CH₂CH₂OOCCH₃. (Not the same as sec-phenylethyl acetate).

Properties: Colorless liquid; peach-like odor. Soluble in alcohol, ether, and most fixed oils. Sp. gr. 1.030-1.034; refractive index 1.497-1.501 (20°C); b.p. 226°C; flash point 230°F; combustible. Toxicity unknown

Derivation: (a) Interaction of ethyl acetate and aluminumphenyl ethylate (b) Interaction of acetic anhydride and phenylethyl alcohol in the presence of sodium acetate.

Grades: Technical; F C.C. Containers: Glass bottles.

Use: Perfumery; laboratory reagent.

sec-phenylethyl acetate. See alpha-methylbenzyl acetate.

phenylethylacetic acid (2-phenylbutyric acid)

C₂H₃CHC₆H₃COOH.
Properties: White crystals with aromatic odor; m.p. 41.0°C (min), insoluble in water; soluble in alcohol, ketones, and esters. Combustible; toxicity unknown. Use: Organic synthesis; laboratory reagent.

2-phenylethyl alcohol. See phenethyl alcohol.

2-phenylethylamine (phenethylamine; 1-amino-2-phenylethane) C₀H₅C₂NH₂.

Properties: Liquid with a fishy odor; absorbs carbon dioxide from the air; strong base; sp. gr. 0.9640; b.p. 194.5°C; soluble in water, alcohol, and ether. Derivation: From phenylethyl alcohol and ammonia under pressure.

Grades: Technical; scintillation.

Containers: Drums.

Hazard: Skin irritant.

Uses: Organic synthesis; laboratory reagent; scintillation counter CO₂ absorber).

2-phenylethyl anthranilate (phenethyl anthranilate) H₂NC₆H₄COOC₂H₄C₆H₅.

Properties: Colorless liquid which yellows with age; odor of grape and orange; sp. gr. 1.14 (25/25°C). Combustible; nontoxic. Uses: Perfume, flavoring.

phenylethyl carbinol. See phenylpropyl alcohol.

phenylethylene. See styrene.

N-phenylethylethanolamine $C_6H_5N(C_2H_5)C_2H_4OH$. Properties: Solid; m.p. 37.2°C; b.p. 268°C (740 mm); sp. gr. 1.04 (20/20°C); very slightly soluble in water. Flash point 270°F (COC). Soluble in alcohol, acetone, benzene. Combustible. Low toxicity. Containers: Drums.

Uses: Solvents; chemical intermediates; preparation of dyes for acetate rayons; laboratory reagent.

pehnyl ethyl ether. See phenetole.

5-phenyl-5-ethylhydantoin (C₆H₅)(C₂H₅)CNHCONHCO

Properties: Colorless, odorless crystalline powder; m.p. 199°C; insoluble in water. Use: Medicine

2-phenylethyl isobutyrate (phenethyl isobutyrate) (CH₃)₂CHCOOC₂H₄C₆H₅.

Properties: Colorless liquid; pleasant odor; sp. gr. 0.988 (25/25°C); refractive index (n 20/D) 1.488; soluble in alcohol and ether. Combustible, nontoxic.

Uses: Perfumes; flavoring

phenylethylmalonylurea. See phenobarbital.

2-phenylethyl mercaptan C₆H₅CH₂CH₂SH

Properties: Liquid. Boiling range 193-225°C; unpleasant odor; sp. gr. 1.0264 (60/60°F); refractive index 1.5582 (n. 20/D); flash point 160°F. Combustible. Containers: Bottles.

Hazard: Probably toxic.

Uses: Organic synthesis; laboratory reagent.

2-phenylethyl phenylacetate (phenethyl phenylacetate)

C₆H₅(CH₂)₂OOCCH₂C₆H₅.
Properties: White crystals; hyacinth odor. Sp. gr. 1.080-1.082; congealing point 27°C. Combustible; low toxicity.

Containers: Bottles. Uses: Perfumery; flavors.

2-phenylethyl propionate (phenethyl propionate) C₂H₅COOC₂H₄C₆H₅.

Properties: Synthetic colorless liquid; flower-fruit odor; miscible with alcohols and ether; sp. gr. 1.012 (25/25°C). Combustible; low toxicity. Uses: Perfumes; flavors.

2-phenylethyl salicylate (phenethyl salicylate) $C_6H_5C_2H_4OOCC_6H_4OH_1$

Properties: Snow-white crystals; very faint aromatic odor. Soluble in 14 parts of 95% alcohol. Congealing point 41.5°C. Combustible. Low toxicity. Uses Flavors.

phenyl ferrocenyl ketone. See benzoylferrocene.

phenyl fluoride. See fluorobenzene.

phenyl fluoromethyl ketone. See fluoroacetophenone.

phenylformamide. See formanilide.

phenylformic acid. See benzoic acid

phenyl gamma acid. See phenyl-2-amino-8-naphthol-6-sulfonic acid.

phenyl glycidyl ether (1,2-epoxy-3-phenoxypropane; PGE) H₂COCHCH₂OC₆H₅.

Properties: Colorless liquid; sp. gr. 1.11; b.p. 245°C; m.p. 3.5°C

Hazard: Toxic by skin absorption; moderately irritating to eyes and skin. Tolerance, 10 ppm in air.

D(-)-alpha-phenylglycine C₆H₅CH(NH₂)COOH Properties: Crystals; m.p. 245-248°C; insoluble in water, ether, alcohol; soluble in acid. Containers: Fiber drums.

Use: Intermediate.

phenylglycolic acid. See mandelic acid.

phenylhydrazine C₆H₅NHNH₂.

Properties: Pale yellow crystals or oily liquid; becomes red-brown on exposure to air. Soluble in alcohol, ether, chloroform, benzene, and dilute acids. Soluble in water, alcohol, and benzene. Sp. gr. 1.0978; m.p. 19.35°C; b.p. 243.5°C, with decomposition. Flash point 192°F (c.c.). Combustible. Autoignition temp. 345°F. Also available as the hydrochloride

Derivation: Reduction of diazotized aniline; followed by reaction with sodium hydroxide.

Grades: Commercial; C.P.; reagent.

Containers: Glass bottles; drums.

Hazard: Highly toxic by inhalation, ingestion, and skin absorption. Tolerance, 5 ppm in air.

WEST

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USPT,JPAB,EPAB,DWPI	17 and (pest\$8)	0	<u>L9</u>
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USPT,JPAB,EPAB,DWPI	13 and (contact\$3 or fumig\$5 or spray\$3)	0	<u>L4</u>
USPT,JPAB,EPAB,DWPI	11 near9 (mite or miticide or miticidal)	3	<u>L3</u>
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1.3 QUE L2 AND (FUMIG) OR CONTACT) OR SPRAYOOR PEST)

FILE BIOSIS, CAPLUS, WPIDS, CROPU ENTERED AT 16 NI 20 ON 03 AUG 2001

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